

EGT3  
ENGINEERING TRIPOS PART IIB

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Wednesday 27 April 2022 2 to 3.40

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**Module 4I10**

**NUCLEAR REACTOR ENGINEERING**

*Answer not more than **three** questions.*

*All questions carry the same number of marks.*

*The **approximate** percentage of marks allocated to each part of a question is indicated in the right margin.*

*Write your candidate number **not** your name on the cover sheet.*

**STATIONERY REQUIREMENTS**

Single-sided script paper

**SPECIAL REQUIREMENTS TO BE SUPPLIED FOR THIS EXAM**

CUED approved calculator allowed

Nuclear Energy Data Book (21 pages)

Engineering Data Books

**10 minutes reading time is allowed for this paper at the start of the exam.**

**You may not start to read the questions printed on the subsequent pages of this question paper until instructed to do so.**

**You may not remove any stationery from the Examination Room.**

1 The axial distribution of neutron flux in a cylindrical PWR fuel pin with pellet radius  $R_{fo}$  is given by  $\phi(z) = \phi_0 \cos(\pi z/L)$  where  $L$  is the active core height. The fuel macroscopic fission cross section is  $\Sigma_f$  and the energy released per fission is  $E_f$ . Most of the released energy is deposited in the fuel at the fission site. However, a fraction,  $\alpha$ , of the total fission energy is deposited directly and axially uniformly in the cladding, while a fraction,  $\beta$ , is deposited directly and axially uniformly in the water coolant. The fuel cladding outer radius is  $R_{co}$  and all other symbols in what follows have their usual meaning.

(a) Show that the axial distribution of bulk coolant temperature is given by

$$T_{\text{coolant}}(z) = T_{\text{in}} + \frac{R_{fo}^2 E_f \Sigma_f \phi_0}{\dot{m} c_p} \left[ (1 - \alpha - \beta)L \left( 1 + \sin \frac{\pi z}{L} \right) + 2(\alpha + \beta) \left( \frac{L}{2} + z \right) \right]$$

[50%]

(b) State the equations along with the boundary conditions necessary to derive the radial temperature distribution within the cladding,  $T_{\text{clad}}(r, z)$ , at axial elevation  $z$ . Explain the physical meaning of each equation. Assume that the cladding thermal conductivity is constant and that thermal contact between the pellet and the cladding is perfect (i.e. there is no temperature drop across the gap).

[30%]

(c) Suggest fuel design options for reducing axial power peaking.

[20%]

2 Reactor designers are considering three alternative coolants with identical properties except that Coolant 2 has 20% higher thermal conductivity than Coolant 1, while Coolant 3 has 20% higher specific heat capacity than Coolant 1 (as shown in Table 1 below). The core power distribution can be assumed to be uniform. The core power is also known to be limited by the maximum fuel temperature of 1000K. The core operating conditions are such that boiling of the coolant is not expected. The coolant inlet temperature is fixed at 500K. The coolant outlet temperature is 550K and the temperature drop across the coolant laminar boundary layer is 20K when Coolant 1 is used.

In answering the following questions, clearly state any assumptions you are making.

- (a) Estimate by what factor the core power can be increased if Coolant 2 is used instead of Coolant 1. [35%]
- (b) Estimate by what factor the core power can be increased if Coolant 3 is used instead of Coolant 1. [35%]
- (c) Discuss whether Coolant 2 or Coolant 3 would offer greater benefit with regards to improving the thermodynamic efficiency of the power conversion cycle. [30%]

Table 1

	Coolant 1	Coolant 2	Coolant 3
Thermal conductivity	$k$	$1.2k$	$k$
Specific heat capacity	$c_p$	$c_p$	$1.2c_p$
Dynamic viscosity	$\mu$	$\mu$	$\mu$
Density	$\rho$	$\rho$	$\rho$

3 Heat from a reactor core is removed by a liquid coolant which also serves as a moderator. The heat is transported by forced circulation to a heat exchanger with a power conversion cycle, as shown in Fig. 1 below. It is proposed to increase the mass flow rate of the coolant in the primary circuit. The thermal power generated by the core, the geometry of all the circuit components and the core outlet coolant temperature are to remain the same.

- (a) List and briefly explain all the positive effects that this change would have on the economics of the power plant. [35%]
- (b) List and briefly explain all the negative effects that this change would have on the economics of the power plant. [35%]
- (c) Propose changes in the core design which can help mitigate the negative effects that you have listed and identify the main drawbacks of these changes. [30%]

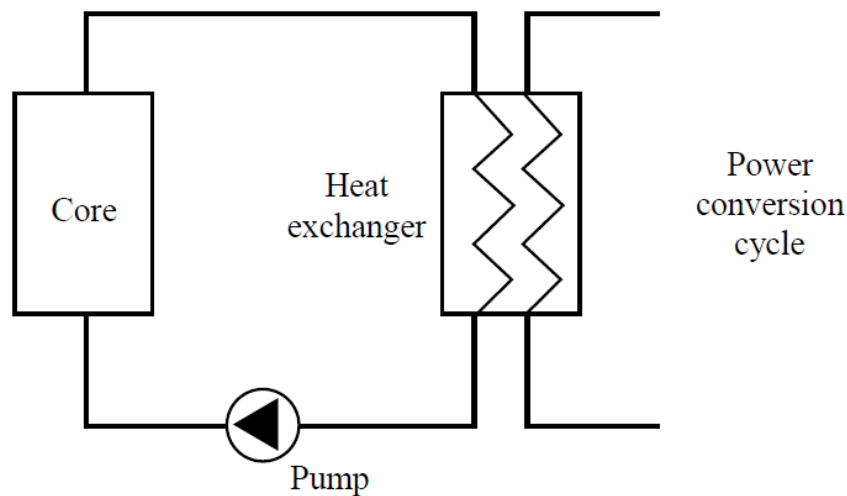


Fig. 1

4 Table 2 below lists three reactivity measurements of a single-batch core at different operating conditions. In answering the following questions, assume that soluble boron reactivity worth is independent of boron concentration and burnup, all Linear Reactivity Model assumptions are valid, and the effects of neutron leakage can be neglected.

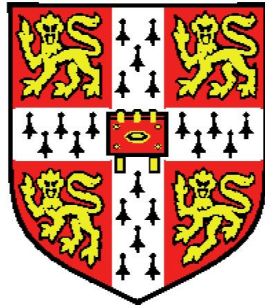
- (a) Estimate the soluble boron reactivity worth. [25%]
- (b) If the same fuel is used in a 4-batch core, estimate the equilibrium cycle burnup and discharge fuel burnup. [25%]
- (c) If the same fuel is used in a 4-batch core, estimate the core critical boron concentration at the beginning of an equilibrium cycle. [25%]
- (d) Discuss the validity of the assumptions of boron reactivity worth being independent of boron concentration and burnup. How would your answers to (a), (b) and (c) change if more realistic boron reactivity worth dependence on boron concentration and burnup were assumed? [25%]

Table 2

Reactivity (pcm)	Burnup (MWd/kg)	Boron Concentration (ppm)
0	0	1800
12000	10	0
6000	30	0

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**Nuclear Energy Data Book**  
**Reactor Physics and Reactor Engineering**

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## 1. NE1: Reactor Physics

### General Data

Speed of light in vacuum	$c$	$299.792458 \times 10^6 \text{ m s}^{-1}$
Avogadro's number	$L$	$6.022141 \times 10^{23} \text{ mol}^{-1}$
Magnetic permeability in vacuum	$\mu_0$	$4\pi \times 10^{-7} \text{ H m}^{-1}$
Planck constant	$h$	$6.626176 \times 10^{-32} \text{ J s}$
Boltzmann constant	$k$	$1.380662 \times 10^{-23} \text{ J K}^{-1}$
Elementary charge	$e$	$1.6021892 \times 10^{-19} \text{ C}$

### Definitions

Unified atomic mass constant	$u$	$1.6605655 \times 10^{-27} \text{ kg}$ (931.5016 MeV)
Electron volt	eV	$1.6021892 \times 10^{-19} \text{ J}$
Curie Ci		$3.7 \times 10^{10} \text{ Bq}$
Barn b		$10^{-28} \text{ m}^2$

### Nomenclature

#### Symbols

$\phi$	Flux distribution (in space)
$\varphi$	Flux distribution (in energy)
$\sigma$	Microscopic cross-section
$\Sigma$	Macroscopic cross-section
$e$	Enrichment
$D$	Diffusion coefficient
$N$	Number of atoms per unit volume
$V$	Volume

#### Superscripts

F	Fuel
M	Moderator

#### Subscripts

a	Absorption
c	Capture
C	Coolant
f	Fission
fe	Fertile
s	Scattering
F	Fuel
$F$	Fast (energy range)
$I$	Intermediate/resonance (energy range)
M	Moderator
$T$	Thermal (energy range)

### Densities and Mean Atomic Weights

	"Nuclear" graphite	Aluminium Al	Cadmium Cd	Gold Au	Uranium U
Density / $\text{kg m}^{-3}$	1600	2700	8600	19000	18900
Atomic weight	12	27	112.4	196	238

**Thermal Neutron Cross-sections (in barns)**

	“Nuclear” graphite	$^{16}_8\text{O}$	$^{113}_{48}\text{Cd}$	$^{235}_{92}\text{U}$	$^{238}_{92}\text{U}$	$^1_1\text{H}$ unbound
Fission	0	0	0	580	0	0
Capture	$4 \times 10^{-3}$	$10^{-4}$	$27 \times 10^3$	107	2.75	0.332
Elastic scatter	4.7	4.2		10	8.3	38

To a first approximation the energy spectrum within the thermal range can be assumed to be proportional to the Maxwell-Boltzmann distribution:

$$\phi(E) \propto \phi_M(E) = \frac{1}{(kT)^2} E \exp(-E/kT)$$

where  $T$  = the temperature (in K)

**Fission Energy**

Kinetic energy of fission fragments	$167 \pm 5 \text{ MeV}$
Prompt $\gamma$ -rays	$6 \pm 1 \text{ MeV}$
Kinetic energy of neutrons	$5 \text{ MeV}$
Decay of fission products $\beta$	$8 \pm 1.5 \text{ MeV}$
$\gamma$	$6 \pm 1 \text{ MeV}$
Neutrinos (not recoverable)	$12 \pm 2.5 \text{ MeV}$
Total energy per fission	$204 \pm 7 \text{ MeV}$

Subtract neutrino energy and add neutron capture energy  $\Rightarrow \sim 200 \text{ MeV / fission}$

**Fission Product Build-up and Decay**

Using a lumped model, the concentration of a fission product will vary over time according to:

$$\frac{dN}{dt} = \gamma_{FP} \Sigma_f \phi - \lambda N - \sigma_a N \phi$$

where  $N$  = the concentration of the product in question

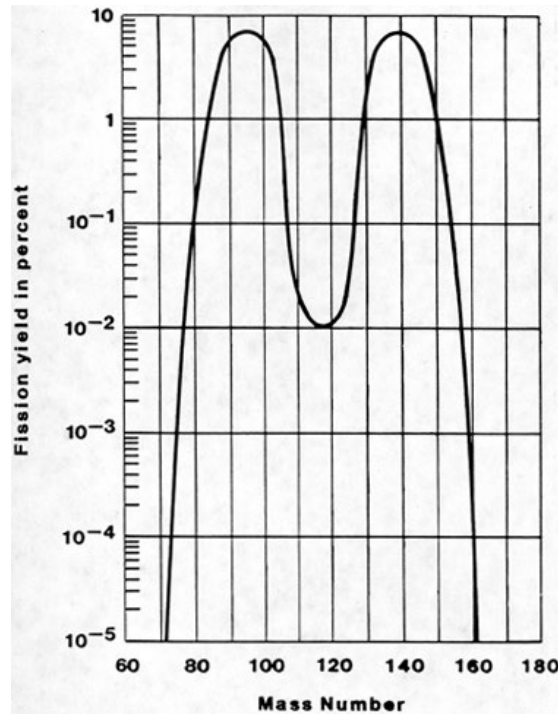
$\gamma_{FP}$  = the *fission yield* (the proportion of reactions producing the product in question)

$\lambda$  = the isotope’s decay constant

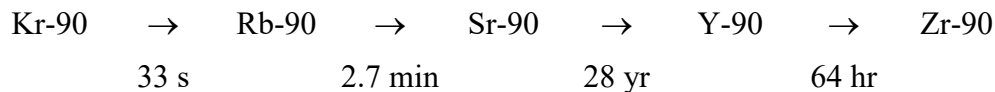
$\sigma_a$  = the isotope’s microscopic absorption cross-section

### Fission Product Yield

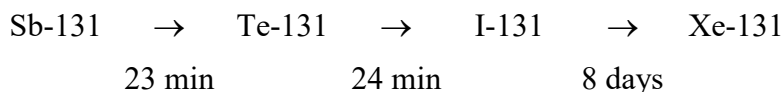
Nuclei with mass numbers from 72 to 158 have been identified, but the most probable split is unsymmetrical, into a nucleus with a mass number of about 138 and a second nucleus that has a mass number between about 95 and 99, depending on the target.



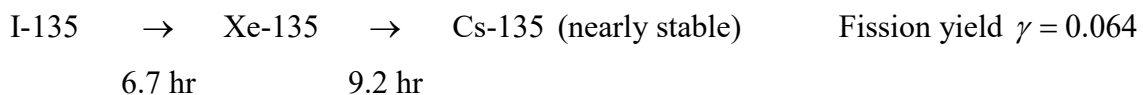
The primary fission products decay by  $\beta^-$  emission. Some important decay chains (with relevant half-lives) from thermal-neutron fission of U-235 are:



Sr-90 is a serious health hazard, because it is bone-seeking.



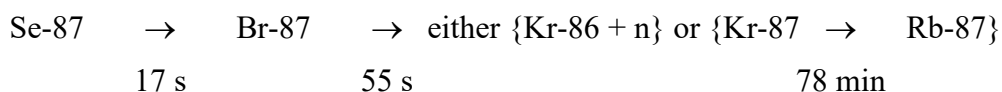
I-131 is a short-lived health hazard. It is thyroid-seeking.



Xe-135 is a strong absorber of thermal neutrons, with  $\sigma_a = 2.65 \text{ Mb}$ .



Sm-149 is a strong absorber of thermal neutrons, with  $\sigma_a = 41 \text{ kb}$ .



This chain leads to a “delayed neutron”.

## Neutrons

Most neutrons are emitted within  $10^{-13}$  s of fission, but some are only emitted when certain fission products, e.g. Br-87, decay.

The total yield of neutrons depends on the target and on the energy of the incident neutron. Some key values are:

Target nucleus	Fission induced by			
	Thermal neutron		Fast neutron	
	$\bar{\nu}$	$\eta$	$\bar{\nu}$	$\eta$
U-233	2.50	2.29	2.70	2.45
U-235	2.43	2.07	2.65	2.30
U-238	—	—	2.55	2.25
Pu-239	2.89	2.08	3.00	2.70

$\nu$  = average number of neutrons emitted per fission

$\eta$  = average number of neutrons emitted per neutron absorbed

### The Four Factor Formula

For a thermal reactor the *infinite multiplication factor*  $k_\infty$  is given by

$$k_\infty = \varepsilon p f \eta_T$$

where  $\varepsilon$  = the fast fission factor

$p$  = the resonance escape probability

$f$  = the thermal utilisation

$\eta_T$  = the number of fission neutrons produced per thermal neutron absorbed in the fuel

$$p = 1 - \frac{V_F N_{fe}}{V_M \xi^M \Sigma_s^M} I$$

where  $I$  = the *resonance integral* defined by

$$I = \int_I \frac{\sigma_a^{fe}(E) \varphi_F(E)}{E \varphi_M(E)} dE$$

$$f = \frac{1}{1 + \zeta \left( V_M \bar{\Sigma}_{aT}^M / V_F \bar{\Sigma}_{aT}^F \right)}$$

where  $\zeta$  = the *thermal disadvantage factor* defined by

$$\zeta = \frac{\bar{\varphi}_{MT}}{\bar{\varphi}_{FT}}$$

### Reactivity Coefficients

*Fuel Temperature Coefficient*  $\alpha_F = \frac{1}{k_\infty} \frac{\partial k_\infty}{\partial \bar{T}_F}$

$\alpha_F$  can be approximated as  $\alpha_F = \frac{1}{\rho} \frac{\partial \rho}{\partial \bar{T}_F} = -\frac{1}{\rho} \frac{V_F N_{fe}}{V_M \xi^M \Sigma_s^M} \frac{\partial I}{\partial \bar{T}_F}$

$\alpha_F$  is also known as the *prompt coefficient*.

*Moderator Temperature Coefficient*  $\alpha_M = \frac{1}{k_\infty} \frac{\partial k_\infty}{\partial \bar{T}_M}$

To a good approximation  $\alpha_M = -\beta_M \left[ \left( \frac{1}{\rho} - 1 \right) - (1 - f) \right]$

where  $\beta_M$  is the volumetric coefficient of thermal expansion at constant pressure given by

$$\beta_M = -\frac{1}{N_M} \frac{\partial N_M}{\partial \bar{T}_M}$$

### Thermal Transients

An approximate lumped parameter model for thermal transients is

$$M_F c_F \frac{d\bar{T}_F}{dt} = P - \frac{1}{R_F} [\bar{T}_F - \bar{T}_C]$$

$$M_C c_C \frac{d\bar{T}_C}{dt} = \frac{1}{R_F} [\bar{T}_F - \bar{T}_C] - 2W_C c_C [\bar{T}_C - T_i]$$

where  $M_F$  = the total fuel mass

$M_C$  = the mass of coolant within the core

$c_F$  = the fuel specific heat capacity

$c_C$  = the coolant specific heat capacity

$\bar{T}_F$  = the volume-average fuel temperature

$\bar{T}_C$  = the volume-average coolant temperature

$T_i$  = the coolant inlet temperature

$P$  = the total power generated

$R_F$  = the overall thermal resistance

$W_C$  = the total coolant mass flow rate through the core

### Composite Reactivity Coefficients

*Isothermal Temperature Coefficient*  $\alpha_T \equiv \frac{d\rho_{fb}}{dT}$

$$\alpha_T = \alpha_F + \alpha_C$$

where  $\alpha_C$  is the *coolant temperature coefficient*

*Power Coefficient*  $\alpha_P \equiv \frac{d\rho_{fb}}{dP}$

$$\alpha_P = R_F \alpha_F + \frac{1}{2W_C c_C} (\alpha_F + \alpha_C)$$

**Nuclear Reactor Kinetics**

Name	Symbol	Concept
Effective multiplication factor	$k_{eff}$	$\frac{\text{production}}{\text{removal}} = \frac{P}{R}$
Excess multiplication factor	$k_{ex}$	$\frac{P - R}{R} = k_{eff} - 1$
Reactivity	$\rho$	$\frac{P - R}{P} = \frac{k_{ex}}{k_{eff}}$
Lifetime	$l$	$\frac{1}{R}$
Reproduction time	$\Lambda$	$\frac{1}{P}$

**Lumped Reactor Kinetics Equations**

$$\frac{dn}{dt} = S + \frac{\rho - \beta}{\Lambda} n + \sum_{i=1}^6 \lambda_i c_i$$

$$\frac{dc_i}{dt} = \frac{\beta_i}{\Lambda} n - \lambda_i c_i$$

where  $n$  = neutron population

$c_i$  = precursor population in group  $i$

$\beta$  = delayed neutron precursor fraction =  $\sum_{i=1}^6 \beta_i$

$\lambda_i$  = decay constant for precursors in group  $i$

$S$  = independent source rate

**Delayed Neutrons**

A reasonable approximation for thermal-neutron fission of U-235 is:

Precursor half-life, s	55	22	5.6	2.1	0.45	0.15	Total
Mean life time of precursor ( $1/\lambda_i$ ), s	80	32	8.0	3.1	0.65	0.22	
Number of neutrons produced per 100 fission neutrons ( $100 \beta_i$ )	0.03	0.18	0.22	0.23	0.07	0.02	0.75

**Neutron Diffusion Equation**

$$\frac{dn}{dt} = -\nabla \cdot \underline{j} + (\eta - 1)\Sigma_a \phi + S$$

where  $\underline{j} = -D \nabla \phi$  (Fick's Law)

$$D = \frac{1}{3 \Sigma_s (1 - \bar{\mu})}$$

with  $\bar{\mu}$  = the mean cosine of the angle of scattering

**Laplacian  $\nabla^2$**

Slab geometry: 
$$\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2}$$

Cylindrical geometry: 
$$\frac{1}{r} \frac{\partial}{\partial r} \left( r \frac{\partial}{\partial r} \right) + \frac{1}{r^2} \frac{\partial^2}{\partial \theta^2} + \frac{\partial^2}{\partial z^2}$$

Spherical geometry: 
$$\frac{1}{r^2} \frac{\partial}{\partial r} \left( r^2 \frac{\partial}{\partial r} \right) + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left( \sin \theta \frac{\partial}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2}{\partial \psi^2}$$

**Bessel's Equation of 0<sup>th</sup> Order**

$$\frac{1}{r} \frac{d}{dr} \left( r \frac{dR}{dr} \right) + R = 0$$

Solution is:

$$R(r) = A_1 J_0(r) + A_2 Y_0(r)$$

$$J_0(0) = 1; Y_0(0) = -\infty;$$

The first zero of  $J_0(r)$  is at  $r = 2.405$

$$J_1(2.405) = 0.5183, \text{ where } J_1(r) = \frac{1}{r} \int_0^r x J_0(x) dx.$$

**Diffusion and Slowing Down Properties of Moderators**

Moderator	Density, g cm <sup>-3</sup>	$\Sigma_a$ , cm <sup>-1</sup>	$D$ , cm	$L^2 = D/\Sigma_a$ , cm <sup>2</sup>
Water	1.00	$22 \times 10^{-3}$	0.17	$(2.76)^2$
Heavy Water	1.10	$85 \times 10^{-6}$	0.85	$(100)^2$
Graphite	1.70	$320 \times 10^{-6}$	0.94	$(54)^2$

The *slowing down decrement*  $\xi$  is the most widely used measure of an isotope's ability to slow neutrons down by elastic scattering:

$$\xi = 1 + \frac{\alpha}{1 - \alpha} \ln \alpha$$

where

$$\alpha = \left[ \frac{A - 1}{A + 1} \right]^2$$

with  $A = M/m$  where  $M$  is the mass of the nucleus and  $m$  the mass of the neutron.

The *slowing down power* of a moderator is then  $\xi \Sigma_s$  and the *slowing down ratio* is  $\xi \Sigma_s / \Sigma_c$ .

**Reactivity Worth**

$$\rho = - \frac{\int \phi \delta \Sigma_a \phi' dV}{\int \phi \nu \Sigma_f \phi' dV}$$

where

$\phi$  = flux distribution prior to control poison insertion

$\phi'$  = flux distribution after control poison insertion

$\delta \Sigma_a$  = change in  $\Sigma_a$  due to control poison insertion

**Health Physics:  $\gamma$ -ray Dose Rate**

$$D = \frac{1.6 \times 10^{-13} A \Sigma E_{\gamma} t}{4\pi \rho R^2}$$

where  $D$  = the dose (in Gy)

$A$  = the activity (in Bq)

$\Sigma$  = the absorption cross-section for  $\gamma$ -rays in human tissue

$E_{\gamma}$  = the (total)  $\gamma$  energy (in MeV)

$t$  = the exposure time (in s)

$\rho$  = the density of tissue

$R$  = the distance from the source (in m)

**Atomic Masses and Naturally Occurring Isotopic Abundances (%)**

	electron	0.00055 u	90.80%	$^{20}_{10}\text{Ne}$	19.99244 u
	neutron	1.00867 u	0.26%	$^{21}_{10}\text{Ne}$	20.99385 u
99.985%	$^1_1\text{H}$	1.00783 u	8.94%	$^{22}_{10}\text{Ne}$	21.99138 u
0.015%	$^2_1\text{H}$	2.01410 u	10.1%	$^{25}_{12}\text{Mg}$	24.98584 u
0%	$^3_1\text{H}$	3.01605 u	11.1%	$^{26}_{12}\text{Mg}$	25.98259 u
0.0001%	$^3_2\text{He}$	3.01603 u	0%	$^{32}_{15}\text{P}$	31.97391 u
99.9999%	$^4_2\text{He}$	4.00260 u	96.0%	$^{32}_{16}\text{S}$	31.97207 u
7.5%	$^6_3\text{Li}$	6.01513 u	0%	$^{60}_{27}\text{Co}$	59.93382 u
92.5%	$^7_3\text{Li}$	7.01601 u	26.2%	$^{60}_{28}\text{Ni}$	59.93079 u
0%	$^8_4\text{Be}$	8.00531 u	0%	$^{87}_{35}\text{Br}$	86.92196 u
100%	$^9_4\text{Be}$	9.01219 u	0%	$^{86}_{36}\text{Kr}$	85.91062 u
18.7%	$^{10}_5\text{B}$	10.01294 u	17.5%	$^{87}_{36}\text{Kr}$	86.91337 u
0%	$^{11}_6\text{C}$	11.01143 u	12.3%	$^{113}_{48}\text{Cd}$	112.90461 u
98.89%	$^{12}_6\text{C}$	12.00000 u		$^{226}_{88}\text{Ra}$	226.02536 u
1.11%	$^{13}_6\text{C}$	13.00335 u		$^{230}_{90}\text{Th}$	230.03308 u
0%	$^{13}_7\text{N}$	13.00574 u	0.72%	$^{235}_{92}\text{U}$	235.04393 u
99.63%	$^{14}_7\text{N}$	14.00307 u	0%	$^{236}_{92}\text{U}$	236.04573 u
0%	$^{14}_8\text{O}$	14.00860 u	99.28%	$^{238}_{92}\text{U}$	238.05076 u
99.76%	$^{16}_8\text{O}$	15.99491 u	0%	$^{239}_{92}\text{U}$	239.05432 u
0.04%	$^{17}_8\text{O}$	16.99913 u		$^{239}_{93}\text{Np}$	239.05294 u
0.20%	$^{18}_8\text{O}$	17.99916 u		$^{239}_{94}\text{Pu}$	239.05216 u
				$^{240}_{94}\text{Pu}$	240.05397 u



## 2. Reactor Engineering

### 2.1 Economics

$$\text{Life time Levelised Electricity Generating Cost (EGC)} = \frac{\sum [(I_t + M_t + F_t + W_t) \times (1+r)^{-t}]}{\sum [E_t \times (1+r)^{-t}]}$$

Where:

$I_t$  = Investment in year  $t$

$M_t$  = Operations & Maintenance in year  $t$

$F_t$  = Fuel costs in year  $t$

$W_t$  = Waste & Decommissioning cost in year  $t$

$E_t$  = Energy generated in year  $t$

$r$  = Discount rate (Real i.e. net of inflation)

### 2.2 Power Cycles

#### *Ideal steam power cycle*

Carnot cycle efficiency  $\eta = \Delta T/T$

Where  $T$  is the heat source temperature and  $\Delta T$  the temperature difference over which the cycle operates.

#### *Ideal gas power cycle*

Efficiency  $\eta = 1 - [1/R_p]^{(\gamma-1/\gamma)}$

Where  $R_p$  is the pressure ratio and  $\gamma$  is the ratio of the specific heats of the gas.

### 2.3 Heat Transfer

#### *Coolants*

The key fluid parameters for some useful reactor coolants:

Coolant	Pressure MPa	Temp. Deg C	Density kg/m <sup>3</sup>	Dyn. Visc. kg/s/m × 10 <sup>-3</sup>	Sp. Ht. kJ/kg/K	Cond. kW/m/K
Light Water	8.8	300	714	0.0897	5.76	0.000541
Carbon Dioxide	4	450	29.5	0.030	1.2	0.00007
Helium	4	450	3.1	0.036	5.2	0.000028
Sodium	0.1	550	817	0.23	1.26	0.061
Lead	0.1	500	10400	1.7	0.145	0.018
Bismuth	0.1	500	9700	1.2	0.135	0.015
LBE	0.1	500	10100	1.2	0.142	0.015
Sat Steam	22	360	144.1	0.0275	25.1	0.000168

Coolant heat transport Figure of Merit

$$FoM = (C_p^{2.8} \rho^2) / \mu^{0.2}$$

Coolant	Pressure MPa	Temp Deg C	FoM	Rel Water	Pr	Rel Pr
Light Water	8.8	300	442963961	100	0.96	100
Carbon Dioxide	4	450	11639	0.0026	0.51	53.9
Helium	4	450	11920	0.0027	0.67	70.0
Sodium	0.1	550	6809827	1.54	0.005	0.5
Lead	0.1	500	1737030	0.39	0.01	1.4
Bismuth	0.1	500	1326300	0.30	0.01	1.1
LBE	0.1	500	1656586	0.37	0.01	1.2
Sat Steam	22	360	1407670277	317.8	4.11	430.2

### Fuel channel & element temperatures

Fourier's Law for heat conduction:

$$\bar{q}'' = -k\bar{\nabla}T$$

Heat conduction (Fourier-Biot) equation:

$$\nabla(k\bar{\nabla}T) + q''' = \rho C_p \frac{\partial T}{\partial t}$$

Plain geometry:

Temperature difference across fuel ( $k_F$ ) with half thickness  $a$ , &  $q'''$  volumetric rating

$$T_{Max} - T_{FO} = q''' a^2 / 2k_F$$

Cladding of thickness:  $t$ , & conductivity  $k_C$

$$T_{Ci} - T_{CO} = q''' a t / k_C$$

Cylindrical geometry

For solid fuel pellets radius  $R_{FO}$  the peak fuel temperature is function of linear rating

$$\int_T^{T_{Max}} k_f(T) dT = \frac{q'}{4\pi} \quad q' \text{ linear rating and } q''' \text{ volumetric rating}$$

or for constant fuel conductivity  $k_f$

$$\text{Temperature difference across fuel} \quad T_{Max} - T_{FO} = q''' R_{FO}^2 / 4k_f$$

$$\text{Also} \quad T_{Max} - T_{FO} = q' / (4\pi k_f)$$

Cladding - Temperature difference across clad

$$T_{Ci} - T_{CO} = q' / (2\pi k_C) [\ln(R_{CO} / R_{Ci})]$$

Spherical geometry

For solid fuel:

$$\text{Temperature difference across fuel } T_{Max} - T_{FO} = q''' R_{FO}^2 / (6k_f)$$

Cladding of fuelled region inner radius  $R_1$ , outer radius  $R_2$  and generating total heat  $Q$ .

$$T_{Ci} - T_{CO} = Q [R_2 - R_1] / [4\pi k_C R_1 R_2]$$

For annular, cylindrical fuel elements and chopped cosine flux profile, the maximum fuel temperature at position  $z$  along the channel:

$$T_F(z) = T_{in} + \frac{q'_0}{\rho A V C_p} \frac{L_e}{\pi} \left[ \sin \frac{\pi z}{L_e} + \sin \frac{\pi L}{2L_e} \right] + \left[ \frac{q'_0}{2\pi R_{Co} h_T} + \frac{q'_0}{2\pi k_C} \ln \left( \frac{R_{Co}}{R_{Ci}} \right) + \frac{q'_0}{2\pi R_G h_G} + \frac{q'_0}{4\pi k_f} \right] \cos \left[ \frac{\pi z}{L_e} \right]$$

at  $z_c$  when:  $\tan \left( \frac{\pi z_c}{L_e} \right) = \left[ \frac{L_e}{\rho A V C_p} \right] / \left[ \frac{1}{2R_{Co} h_T} + \frac{1}{2R_C h_G} + \frac{1}{2k_C} \ln \left( \frac{R_{Co}}{R_{Ci}} \right) + \frac{1}{4k_f} \right]$

where $L_e$	effective core height	$q'_0$	peak linear heat rating
$h_T$	coolant heat transfer coefficient	$k_C$	clad thermal conductivity
$h_G$	gas gap heat transfer coefficient	$k_f$	fuel mean thermal conductivity
$\rho$	coolant density	$A$	flow area
$V$	coolant flow velocity	$C_p$	coolant specific heat

**Single phase heat transfer and hydraulics**

There is range of correlations for convective heat transfer to fluids. The most commonly used correlations for forced convection in nuclear energy applications are:

Gas and liquid coolants (excluding liquid metals) - turbulent flow:

$$Nu = 0.023 Re^{0.8} Pr^{0.33} \quad Re > 10,000 \quad 0.7 > Pr > 120$$

Liquid metal coolants – turbulent flow:

$$Nu = 6 + 0.006 Pe \quad \text{square pin array}$$

$$Nu = 0.047(1 - e^{-3.8(P/D-1)})(Pe^{0.77} + 250) \quad \text{triangular pin array}$$

Where both Nusselt and Reynolds numbers are based on hydraulic diameter:

$$Nu = h_r D_h / k \quad Pe = RePr = (\rho V D_h C_p) / k$$

Hydraulic diameter  $D_h = 4 \times \text{Flow Area} / \text{Wetted Perimeter}$

$P/D = \text{Pitch/Diameter of pin array}$

Moody friction factor  $f$

Laminar flow:  $f = \frac{64}{Re}$

Turbulent flow:  $f = \frac{0.184}{Re^{0.2}}$  (smooth pipes)

## 2.4 Two Phase Flow Quantities

Void Fraction  $a = \frac{A}{A_v + A_l}$

Flow based Quality  $X = \frac{\dot{m}_v}{\dot{m}_v + \dot{m}_l}$  Slip ratio  $S = \frac{u_v}{u_l}$

Relationship between Void Fraction and Quality

$$a = \frac{X}{x + \left(\frac{\rho_v}{\rho_l}\right)(1-x)S} \quad \text{where } (v) \text{ denotes vapour \& } (l) \text{ liquid}$$

## 2.5 In-core Fuel Management Equilibrium Cycle Burn-up Length Ratio

For M-batch refuelling:  $\frac{B_M}{B_1} = \frac{2M}{M+1}$

Where:  $B_M$  is the irradiation burn-up for M batch refuelling.

## 2.6 Decay Heat Estimation

Decay Heat (Beta & Gamma):  $P(t) = 0.066 P_0 (t^{-0.2} - (t + t_0)^{-0.2})$

where  $P_d(t)$  = the power generation due to  $\beta$  decay and  $\gamma$ -radiation

$P_0$  = the power before shutdown,

$t_0$  = the time of operation before shutdown (in s)

$t$  = the elapsed time since shutdown (in s)

**2.7 Reactor Materials Properties**

Material	Temp, K	Density, g/cm <sup>3</sup>	Linear expansion, K×10 <sup>-6</sup>	Specific Heat, kJ/kg/K	Thermal Conduct, W/m/K	UTS, MPa	Yield, MPa	Young's Modulus, GPa
<b>Graphite Nuclear grade</b>	300	1.7	-	0.71	156	13.8	58(C)	-
	500		3.6	1.25	118	15.9		9.0
	800		5.0	1.67	73	17.2		
	1400		6.3	1.88	36	19.3		
	2500		8.5	-	31	27.6		
<b>Steel, carbon A 533B</b>	300	7.86	-	0.50	52	500	340	207
	600		10.2	0.59	43	530	280	182
	750		10.4	0.63	38	450	240	172
	800		10.4	0.67	35	390	200	169
<b>Steel, stainless type 347</b>	300	7.95	-	0.50	14	520	210	-
	500	7.86	16.9	0.52	17	420	-	173
	700	7.71	17.4	0.55	20	400	150	166
	800	-	18.5	0.57	22	390	-	157
<b>Uranium Carbide</b>	300	13.63	-	0.15	-	-	372(C)	214
	800		-		23			
	1250		10.3		23			
<b>Uranium Dioxide</b>	300	10.98	-	0.23	8.0	960(C)		183
	500		9.0	0.28	6.1			-
	800		10.1	0.30	4.1			165
	110		-	0.31	2.6			
	1400		12.8	0.32	2.2			
	2300			0.42	2.3			
<b>Zircalloy-2</b>	300	6.56	-	0.28	12.7	490	300	95
	500		-	0.31	15.2	280	170	90
	600		6.5	0.33	16.5	210	117	78
	800			0.35	18.9	-		
	1000			0.37	21.6	-		

**Source: Glasstone & Sesonske**

### 3. Fuel Cycle, Waste & Decommissioning

#### 3.1 Enrichment of Isotopes

Separative Capacity D:

$$D = W(2x_w - 1) \ln \frac{x_w}{1 - x_w} + P(2x_p - 1) \ln \frac{x_p}{1 - x_p} - F(2x_f - 1) \ln \frac{x_f}{1 - x_f}$$

Where  $x_w$  is the mol fraction of the tails

$x_p$  is the mol fraction of the product

$x_f$  is the mol fraction of the feed

$W$  is flow rate of tails (mols/s)

$P$  is flow rate of product

$F$  is flow rate of feed

Separative Work:

$$S = E_w(2x_w - 1) \ln \frac{x_w}{1 - x_w} + E_p(2x_p - 1) \ln \frac{x_p}{1 - x_p} - E_f(2x_f - 1) \ln \frac{x_f}{1 - x_f}$$

where  $E_w$  is mass of tails (kg or Te)

$E_p$  is mass of product

$E_f$  is mass of feed

All fractions are mass fractions

Value function:  $v(x) = (2x - 1) \ln \left( \frac{x}{1 - x} \right) \approx -\ln(x)$  for small  $x$

where  $x$  is the mol or mass fraction of the component in question.

#### 3.2 Hold up and decay (Extended Bateman's Equation)

$$N_i = \lambda_1 \lambda_2 \dots \lambda_{i-1} P \sum_{j=1}^i \frac{[1 - \exp(-\lambda_j T)] \exp(-\lambda_j \tau)}{\lambda_j \prod_{\substack{k=1 \\ k \neq j}}^i (\lambda_k - \lambda_j)}$$

where  $N_i$  = number of atoms of nuclide  $I$  at time  $T + \tau$ ,  $T$  = filling time,

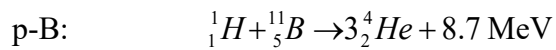
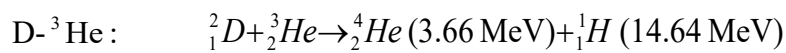
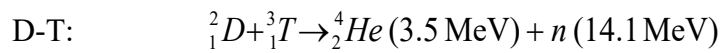
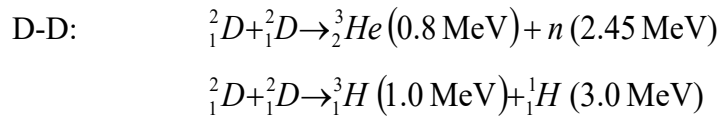
$\lambda_j$  = decay constant of nuclide  $j$ ,  $\tau$  = decay hold-up time after filling,

$P$  = parent nuclide arising rate (atoms/unit time)

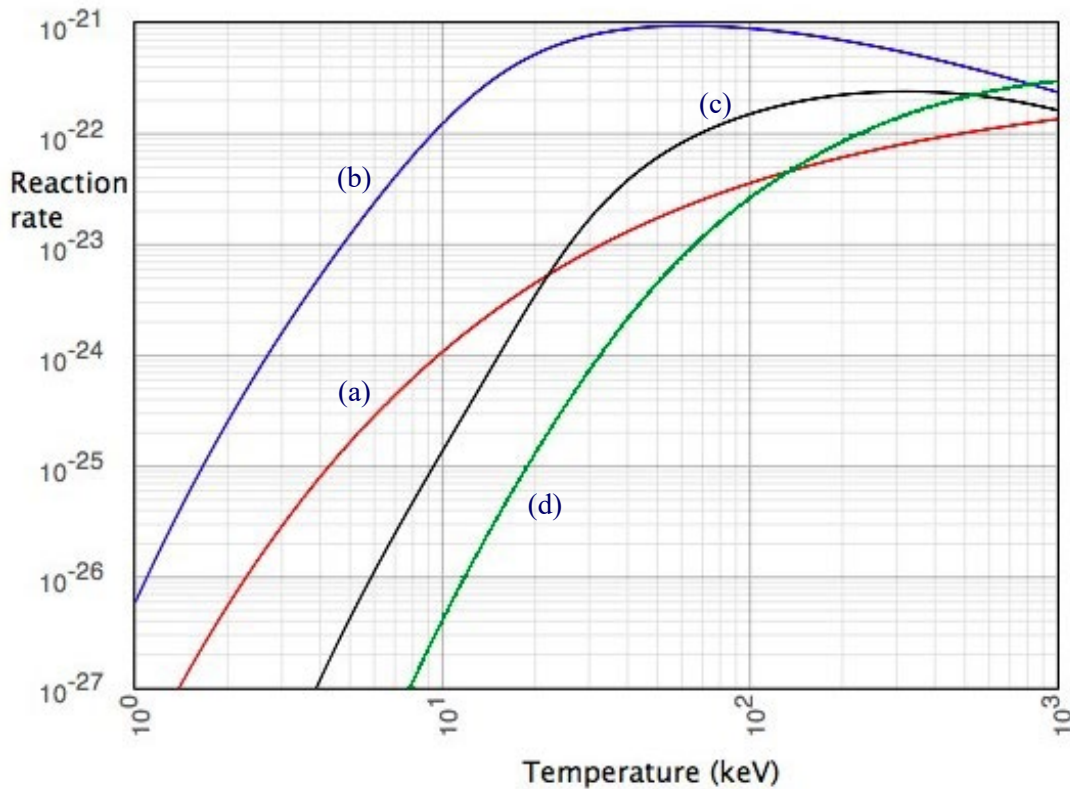
Note: decay constant  $\lambda_j$  and filling time  $P$  must have the same units of time.

## 4. Fusion

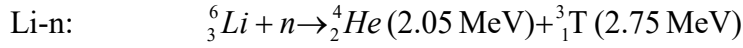
### 4.1 Major fusion reactions:



The reaction rate  $\langle \sigma v \rangle (\text{m}^3 \text{s}^{-1})$  for these reactions is shown as a function of plasma temperature in keV in the graph below. (a) D-D (including both reactions); (b) D-T; (c) D-<sup>3</sup>He; (d) p-B.



#### 4.2 Tritium breeding reaction:



#### 4.3 Plasma parameters:

$$\text{Plasma } \beta: \quad \beta = \frac{\langle P \rangle}{\frac{B^2}{2\mu_0}}$$

$$\text{Normalised } \beta: \quad \beta_N = \beta \frac{aB_T}{I_P} (\%, \text{ m, T, MA})$$

$$\text{Greenwald density limit } n_G: \quad n_G = \frac{I_P}{\pi a^2} (10^{20} \text{ m}^{-3}, \text{ MA, m})$$

$$\text{Plasma parameter } \Lambda_D: \quad \Lambda_D = \frac{4\pi \epsilon_0 T_e^{3/2}}{3 e^3 n_e^{1/2}} \gg 1$$

#### 4.4 Tokamak ELMy H-mode confinement time scaling law IPB98(y,2):

$$\tau_e = H \times 0.145 I^{0.93} R_0^{1.39} a^{0.58} \kappa^{0.78} n_{20}^{0.41} B_0^{0.15} A^{0.19} P^{-0.69} (\text{MA, m, m, } 10^{20} \text{ m}^{-3}, \text{ T, MW})$$

#### 4.5 Lawson criteria:

$$n_e \tau_E \geq \frac{12 k_B T}{E \langle \sigma v \rangle} (\approx 1.5 \times 10^{20} \text{ s m}^{-3} \text{ for D-T})$$

$$n_e T_e \tau_E \geq \frac{12 k_B T^2}{E \langle \sigma v \rangle} (\approx 10^{21} \text{ keV s m}^{-3} \text{ for D-T})$$

#### 4.6 Bremsstrahlung radiation:

$$P_{\text{brem}} \approx 0.85 \text{Pr}^2 \frac{Z_{\text{eff}}^2}{T_{10\text{keV}}^{3/2}} (\text{MW m}^{-3}, \text{ atm, } 10 \text{ keV})$$

$$Z_{\text{eff}} = \frac{\sum Z_i^2 n_i}{\sum Z_i n_i} \quad (\text{Effective plasma ion charge})$$



**4.7 Larmour (gyro) radius:**

$$\rho_L = \frac{v_{th}}{\omega_{cy}} = \frac{v_{th}}{\frac{ZeB}{m}} = \frac{(2mT)^{1/2}}{ZeB}$$

**4.8 Molar masses of common fusion isotopes (and neutrons):**

Isotope	Symbol	Molar mass (g)
Neutron	${}^1_0n$	1.0087
Hydrogen	${}^1_1H$	1.0079
Deuterium	${}^2_1D$	2.0141
Tritium	${}^3_1T$	3.01605
Helium-3	${}^3_2He$	3.01603
Helium	${}^4_2He$	4.0026
<b>Boron-11</b>	${}^{11}_5B$	<b>11.0093</b>

**4.9 Nomenclature**

$\beta$	ratio of thermal/magnetic pressure (%)	$\beta_N$	ratio to Troyon stability limit
$B$	Magnetic field (Tesla)	$I_P$	plasma current (MA)
$E$	Energy released per fusion (MeV )	$\langle\sigma_v\rangle$	average fusion cross section
$n$	number density		
$Pr$	Pressure (either Pa, or bars)		
$T_e$	Temperature in energy units (eV, or keV)		
$Z$	Ion charge number		
$\rho_L$	Lamour radius		
$v_{th}$	thermal velocity	$\omega_{cy}$	cyclotron frequency

## 5. Nuclear Safety

### 5.1 UK NII/ONR system of Safety Categorisation and Classification

A **safety categorisation** scheme could be determined on the following basis:

- a. Category A – any function that plays a principal role in ensuring nuclear safety.
- b. Category B – any function that makes a significant contribution to nuclear safety.
- c. Category C – any other safety function.

The **method** for categorising safety functions should take into account:

- a. the consequence of failing to deliver the safety function;
- b. the extent to which the function is required, either directly or indirectly, to prevent, protect against or mitigate the consequences of initiating faults;
- c. the potential for a functional failure to initiate a fault or exacerbate the consequences of an existing fault;
- d. the likelihood that the function will be called upon.

A safety classification scheme for **safety measures** (structures, systems and components) could be determined on the following basis:

- a. Class 1 – any structure, system or component that forms a principal means of fulfilling a Category A safety function.
- b. Class 2 – any structure, system or component that makes a significant contribution to fulfilling a Category A safety function, or forms a principal means of ensuring a Category B safety function.
- c. Class 3 – any other structure, system or component.

### 5.2 Long term Health Risks related to Radiation

The International Commission for Radiological Protection has recommended that the following expression should be used to assess the risk of death from radiation.

- Risk of death from developing a fatal cancer = Radiation dose(in Sv) x Risk Factor
- They recommend a risk factor for adults of 0.04 per Sievert.

### 5.3 Safety Criteria

#### I. UK Radiation dose limits for normal (non accident) operations

Risk Group	Industrial workers	Members of the general public
Basic Safety Level	20mSv	1mSv
Basic Safety Objective	1mSv	0.02

#### II. UK Risk Criteria for nuclear accidents

Risk Group	Industrial Workers	Member of the general public
Basic Safety Level	$10^{-4}$	$10^{-4}$
Basic Safety Objective	$10^{-6}$	$10^{-6}$

#### III. ALARP cost benefit criteria

- Where the benefit is the prevention of death, the current convention used by UK Health & Safety Executive, when conducting a Cost Benefit Analysis is to adopt a benchmark value of about £1,000,000 (2001 prices) for the **value of preventing a fatality (VPF)**.
- However due to the aversion caused by deaths from cancer and radiation a premium should be applied to this value. **A factor of 2** is recommended to be applied for any deaths caused by radiation and therefore £2m for VPF should be used for nuclear facilities and or radioactive installations.
- The disproportionation factor applies a weighting factor so that more money should be spent to improve the safety of higher risk facilities in comparison to lower risks facilities.
  - The values to be used are at least a factor of 10 if the risks are predicted to be near the **Basic Safety Level (BSL)**, and
  - A factor of 3 if the risks are predicted to be near the **Basic Safety Objective (BSO)**.

#### IV. Dose Staircase Criteria

Targets for the predicted frequency of any single accident, which could give doses to a person on the site, are for:	Predicted frequency per annum BSL/ BSO
Effective dose, mSv	
2 – 20	$1 \times 10^{-1} / 1 \times 10^{-3}$
20 – 200	$1 \times 10^{-2} / 1 \times 10^{-4}$
200 – 2000	$1 \times 10^{-3} / 1 \times 10^{-5}$
> 2000	$1 \times 10^{-4} / 1 \times 10^{-6}$