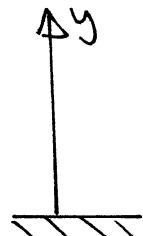


Paper 4 Fluid Mechanics and Heat Transfer, 1996

Solutions

① (i) Hydrostatics

$$\frac{dp}{dy} = -\rho g$$



$$\Rightarrow \frac{dp}{dy} = -\left(\frac{P}{RT_0}\right) g$$

$$\Rightarrow \frac{dp}{P} = -\frac{g}{RT_0} dy$$

$$\Rightarrow \ln\left(\frac{P}{P_0}\right) = -\frac{g}{RT_0} y$$

$$\Rightarrow P = P_0 e^{-\frac{g}{RT_0} y}$$

Expand $e^{-\frac{gy}{RT_0}}$ in Taylor series

$$P = P_0 \left[1 - \frac{g}{RT_0} y + \dots \right]$$

$$\Rightarrow P = P_0 - \frac{P_0}{RT_0} gy + \dots$$

$$\Rightarrow P = P_0 - \underline{\rho_0 gy + \dots}$$

Need only retain first term in expansion if

$$\frac{gy}{RT_0} \ll 1$$

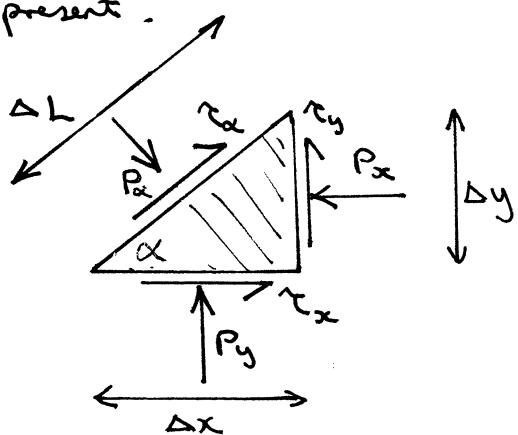
$$\Rightarrow \frac{\rho_0 gy}{\rho_0} \ll 1$$

$$\Rightarrow \underline{y \ll \frac{\rho_0}{\rho_0 g}}$$

(ii) Pascal's law: Pressure same in all directions if the fluid is stationary.

Technical defn. of fluid: Fluid cannot sustain a shear stress and stay at rest.

If fluid is in motion then there will, in general, be shear stresses present.



Forces on element are of order $\Delta y P$, Δx^2 etc.

$$\sum F = \Delta m a$$

But Δm is of order $\Delta x \Delta y$. Thus, for small element

$$\sum F = 0.$$

Resolve in x direction:

$$-P_x \Delta y + \tau_x \Delta x + (P_x \Delta L) \sin \alpha + (\tau_x \Delta L) \cos \alpha = 0$$

$$\Rightarrow -P_x \Delta y + \tau_x \Delta x + P_x \Delta y + \tau_x \Delta x$$

$$\Rightarrow P_x \neq P_x \text{ unless } (\tau_x + \tau_x) = 0$$

② (i)

Bernoulli $1 \rightarrow 2$.

$$\rho_1 + \frac{1}{2} \rho v_1^2 + \rho g(h+w) = \rho_2 + \frac{1}{2} \rho v_2^2 + \rho g(z)$$

$$\rho_1 = \rho_a, \quad \rho_2 = \rho g w + \rho_a$$

$$\Rightarrow \cancel{\frac{1}{2} \rho v_1^2 + \rho g(h+w)} = \cancel{\rho g w} + \frac{1}{2} \rho v_2^2$$

$$\Rightarrow \frac{1}{2} \rho (v_2^2 - v_1^2) = \rho g h$$

$$\Rightarrow v_2^2 \left[1 - \left(\frac{A_2}{A_1} \right)^2 \right] = 2gh$$

$$\Rightarrow v_2^2 \left[1 - \left(\frac{d}{D} \right)^2 \right] = 2gh$$

$$\Rightarrow v_2 = \frac{\sqrt{2gh}}{\sqrt{1 - \left(\frac{d}{D} \right)^2}}$$

But $Q = A_2 v_2 = A_1 \left(-\frac{dh}{dt} \right)$

$$\Rightarrow \frac{dh}{dt} = - \frac{A_2}{A_1} v_2$$

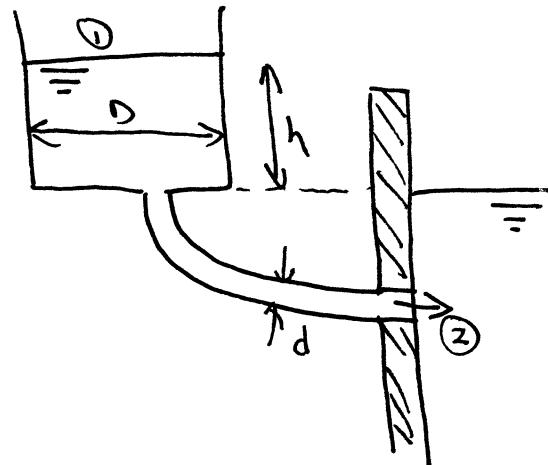
$$\Rightarrow \frac{dh}{dt} = - \left(\frac{d}{D} \right)^2 \sqrt{\frac{2gh}{1 - \left(\frac{d}{D} \right)^2}}$$

Integrate

$$h^{-1/2} dh = - \left(\frac{d}{D} \right)^2 \sqrt{\frac{2g}{1 - \left(\frac{d}{D} \right)^2}} dt$$

$$\Rightarrow 2(\sqrt{h} - \sqrt{h_0}) = - \left(\frac{d}{D} \right)^2 \sqrt{\frac{2g}{1 - \left(\frac{d}{D} \right)^2}} t$$

Let t_{final} be time to drain.



$$\Rightarrow t_{\text{final}} = 2 \sqrt{\frac{1 - (\frac{d}{D})^4}{2g}} \left(\frac{D}{d}\right)^2 \sqrt{h_0}$$

(ii) Extended Bernoulli $1 \rightarrow 2$

$$\underbrace{\left[0 + \frac{1}{2} \rho v_1^2 + \rho g(h+w) \right]}_{\text{Initial Energy}} - \underbrace{\left[\rho g w + \frac{1}{2} \rho v_2^2 + 0 \right]}_{\text{Final Energy}} = \frac{fL}{d} \frac{1}{2} \rho v_2^2$$

$$\Rightarrow \frac{1}{2} \rho v_1^2 = \frac{1}{2} \rho v_2^2 - \frac{fL}{d} \frac{1}{2} \rho v_2^2 + \rho g h = 0$$

$$\Rightarrow v_2^2 \left[1 + \frac{fL}{d} \right] - v_1^2 = 2gh$$

$$\Rightarrow v_2^2 \left[1 + \frac{fL}{d} - \left(\frac{A_2}{A_1} \right)^2 \right] = 2gh$$

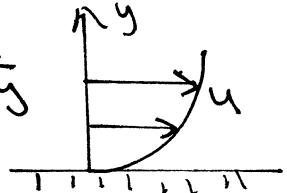
$$\Rightarrow v_2 = \sqrt{\frac{2gh}{1 + \frac{fL}{d} - \left(\frac{d}{D} \right)^4}}$$

Rest of analysis comes over as before but with $1 - \left(\frac{d}{D} \right)^4$ replaced by, $1 + \frac{fL}{d} - \left(\frac{d}{D} \right)^4$.

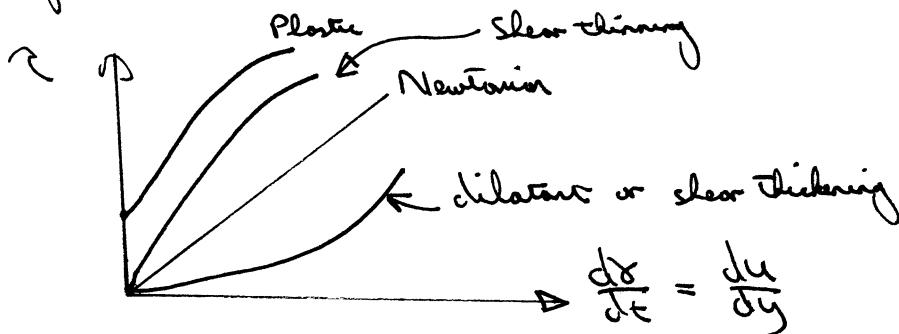
$$t_{\text{final}} = 2 \sqrt{\frac{1 + \frac{fL}{d} - \left(\frac{d}{D} \right)^4}{2g}} \left(\frac{D}{d}\right)^2 \sqrt{h_0}$$

neglect friction if $\frac{fL}{d} \ll 1$.

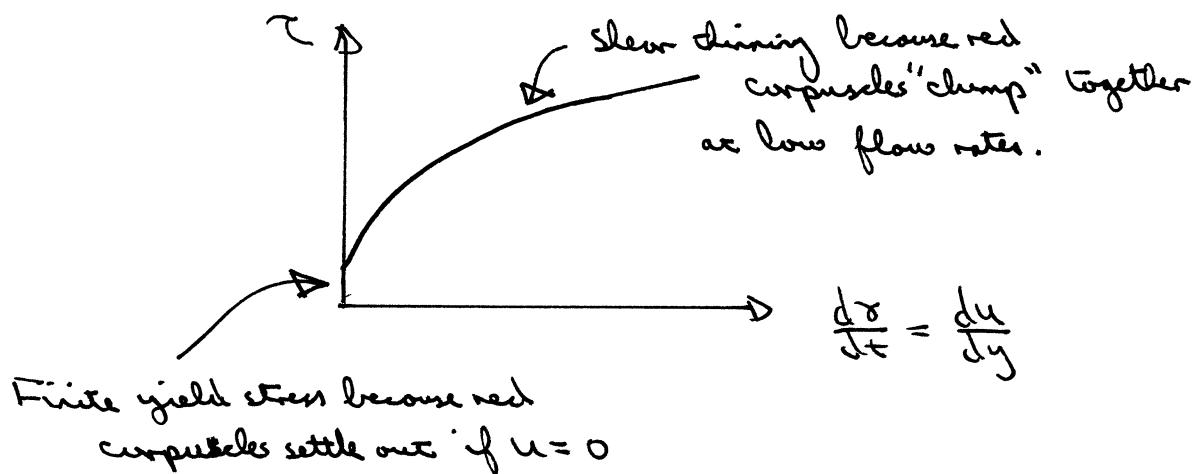
③ (i) Newton's law of viscosity: $\tau = \mu \frac{du}{dy}$



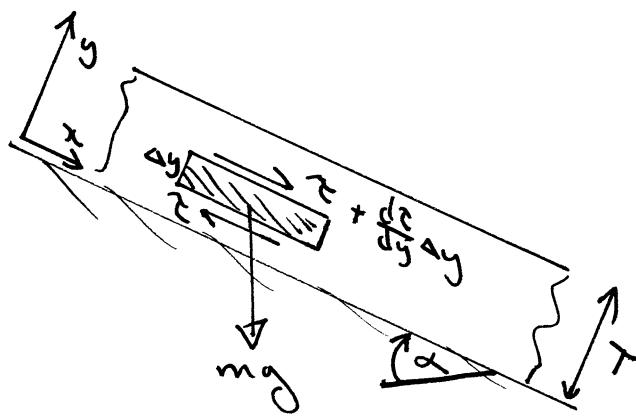
Other types of behavior are:



One non-Newtonian fluid is blood



(ii)



Force balance:

$$x: mg \sin \alpha + \left(\tau + \frac{\partial \tau}{\partial y} dy \right) \Delta x - \tau \Delta x = 0$$

\uparrow
 $\rho \Delta x \Delta y$

$$\Rightarrow \cancel{\rho \Delta x \cancel{dy} g \sin \alpha} + \frac{de}{dy} \cancel{\Delta x \cancel{dy}} = 0$$

$$\Rightarrow \underline{\frac{de}{dy} = -\rho g \sin \alpha}$$

$$e = \mu \frac{du}{dy} \Rightarrow \mu \frac{d^2u}{dy^2} = -\rho g \sin \alpha$$

$$\Rightarrow e = \mu \frac{du}{dy} = -\rho g \sin \alpha (y + C)$$

$$\text{But : } e = 0 \text{ at } y = T \text{ (surface)}$$

$$\Rightarrow e = \rho g \sin \alpha (T - y)$$

$$\Rightarrow \underline{\frac{du}{dy} = \frac{\rho g \sin \alpha}{\mu} (T - y)}$$

$$\Rightarrow u = \frac{\rho g \sin \alpha}{\mu} \left(Ty - \frac{y^2}{2} \right) + C$$

$$\text{But : } u = 0 \text{ at } y = 0 \Rightarrow C = 0$$

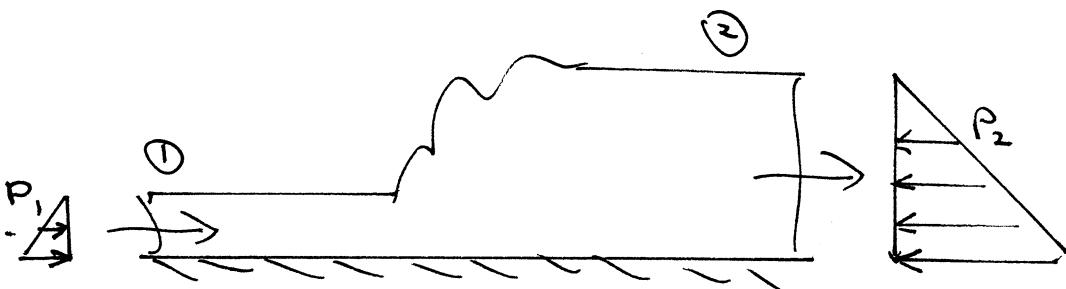
$$\Rightarrow \underline{u = \frac{\rho g \sin \alpha}{\mu} y \left(T - \frac{y}{2} \right)}$$

$$Q = \int_0^T u dy = w \rho \frac{g \sin \alpha}{\mu} \left(\frac{T^2}{2} - \frac{T^3}{6} \right) = \underline{w \rho g \sin \alpha \frac{T^3}{3\mu}}$$

$$\Rightarrow \underline{T = \left(\frac{3\mu Q}{w \rho g \sin \alpha} \right)^{1/3}}$$

$$(iii) \quad T = \left(\frac{3 \times 10^{-6} \times 45 \times 10^6}{0.3 \times 9.81 \times \sin 6^\circ} \right)^{1/3} = \underline{0.760 \text{ mm}}$$

- ④ (i) Pressure distribution is hydrostatic because there is no vertical acceleration in fluid so the vertical force balance on an element of fluid is pressure versus weight i.e. hydrostatic.



$$\sum F_x = \sum_{\text{OUT}} m u_x - \sum_{\text{IN}} m u_{max}$$

$$\int_0^{h_1} \rho_1 dy - \int_0^{h_2} \rho_2 dy = \underbrace{(\rho h_1 v_1)}_{m} v_2 - (\rho h_1 v_1) v_1$$

$$\Rightarrow \frac{1}{2} \cancel{\rho g h_1^2} - \frac{1}{2} \cancel{\rho g h_2^2} = \cancel{\rho h_1 v_1} (v_2 - v_1)$$

$$\underline{\underline{\Rightarrow \frac{1}{2} (h_1^2 - h_2^2) = h_1 v_1 (v_2 - v_1)}}$$

(ii) Need to eliminate v_2 using continuity $h_1 v_1 = h_2 v_2$

$$\underline{\underline{\Rightarrow \frac{1}{2} (h_1^2 - h_2^2) = h_1 v_1^2 \left(\frac{v_2}{v_1} - 1 \right) = h_1 v_1^2 \left(\frac{h_1}{h_2} - 1 \right)}}$$

$$\Rightarrow (h_1^2 - h_2^2) = 2 \frac{v_1^2}{g h_1} \frac{h_1^2}{h_2} (h_1 - h_2)$$

$$\Rightarrow (h_1 + h_2) = 2 \frac{v_1^2}{g h_1} \frac{h_1^2}{h_2}$$

$$\underline{\underline{\Rightarrow \left(\frac{h_2}{h_1} \right) \left(1 + \frac{h_2}{h_1} \right) = 2 \frac{v_1^2}{g h_1}}}$$

$$\text{For } h_1 = 1.5, \quad V_1 = 9.6, \quad 2F_r^2 = 12.53$$

$$\left(\frac{V_2}{V_1}\right)^2 + \frac{h_2}{h_1} - 12.53 = 0$$

$$\Rightarrow \frac{h_2}{h_1} = 3.07 \text{ m}$$

$$\Rightarrow \underline{\underline{h_2 = 4.61 \text{ m}}}$$

(iii) For dynamic similarity, need F_r same in lab test.

$$\Rightarrow \frac{V_{\text{test}}}{\sqrt{h_{\text{test}}}} = \frac{V_{\text{real}}}{\sqrt{h_{\text{real}}}}$$

$$\Rightarrow V_{\text{test}} = V_{\text{real}} \sqrt{\frac{h_{\text{test}}}{h_{\text{real}}}} = 9.6 \sqrt{\frac{1}{3}} \\ = \underline{\underline{5.54 \text{ m/s}}}$$

⑤ (i)

$$T - T_w = (T_o - T_w) \left(1 - \frac{2}{5} \left(\frac{r}{R} \right)^2 + \frac{4}{5} \left(\frac{r}{R} \right)^3 \right)$$

T_o = center-line temp.

$$Nu = \frac{hD}{k}, \quad h = \frac{\dot{Q}}{(T_o - T_w) A} = \frac{-k \left(\frac{\partial T}{\partial r} \right)_w}{T_o - T_w}$$

$$\left(\frac{\partial T}{\partial r} \right)_w = (T_o - T_w) \left[-\frac{18}{5} \frac{1}{R} + \frac{12}{5} \frac{1}{R} \right]$$

$$= -\frac{(T_o - T_w)}{R} \frac{6}{5}$$

$$\Rightarrow h = \left(\frac{-k}{(T_o - T_w)} \right) \left(-\frac{6}{5} \frac{(T_o - T_w)}{R} \right)$$

$$= \frac{6}{5} \frac{k}{R}$$

$$Nu = \frac{hD}{k} = \frac{6}{5} \frac{k}{R} \frac{2R}{R} = \frac{12}{5} = \underline{\underline{2.4}}$$

$$(ii) \quad \frac{\partial T}{\partial r} = (T_o - T_w) \left(-\frac{18}{5} \frac{1}{R^2} + \frac{12}{5} \frac{r^2}{R^3} \right)$$

$$\Rightarrow \frac{\partial T}{\partial r} \Big|_o = 0 \quad \checkmark \quad \text{This must be true to avoid a cusp in the temperature profile i.e. symmetry}$$

$$\frac{\partial}{\partial r} \left(r \frac{\partial T}{\partial r} \right) = \frac{\partial}{\partial r} \left[(T_o - T_w) \left(-\frac{18}{5} \frac{1}{R^2} + \frac{12}{5} \frac{r^2}{R^3} \right) \right]$$

$$= (T_o - T_w) \left(-\frac{36}{5} \frac{1}{R^3} + \frac{36}{5} \frac{r}{R^4} \right)$$

$$\Rightarrow \frac{\partial}{\partial r} \left(2\pi r \frac{\partial T}{\partial r} \right)_w = 0 \quad \checkmark$$

This must hold true because of no-slip at the wall, so no heat is transported axially by fluid, so locally we have a heat balance due to conduction.

$$\dot{Q}_2 = -k \frac{\partial T}{\partial r} \times 2\pi r \quad (\text{per unit length})$$

$$\dot{Q}_1 = -k \frac{\partial T}{\partial r} \times 2\pi r$$

For steady state, $\dot{Q}_1 = \dot{Q}_2$

$$\Rightarrow \cancel{k} \left(2\pi r k \frac{\partial T}{\partial r} \right) = 0$$

(iii)

$$h = f(k, \rho c_p, \nu, \sigma, \bar{\nu})$$

7 independent parameters ; 4 dimension (m, L, T, Θ)

\Rightarrow 3 groups.

They are,

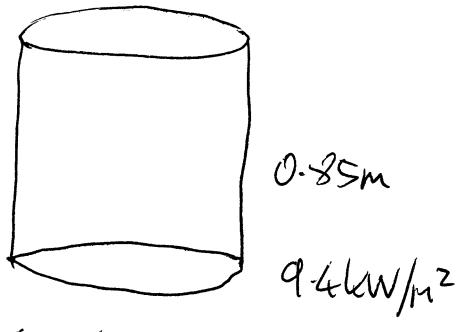
$$\left\{ \begin{array}{l} Nu = \frac{hD}{k} \quad (\text{or Stanton no. will do}) \\ Re = \frac{UD}{\nu} \\ Pr = \frac{(\rho c_p)\nu}{k} \end{array} \right.$$

IB Paper 4 Exam Question Crib 1996

Radiation Heat Transfer and Combustion

$$\varepsilon = 0.63, 500K$$

6.

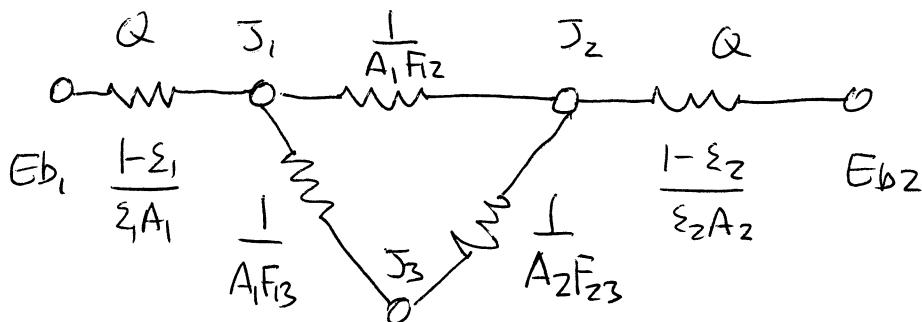


Let 1 = bottom

2 = top

3 = sides

$$\varepsilon = 0.85$$



$$Q = \frac{E_{b1} - E_{b2}}{R_T} ; R_T = \frac{1 - \varepsilon_1}{\varepsilon_1 A_1} + \frac{1 - \varepsilon_2}{\varepsilon_2 A_2}$$

$$\frac{1}{R} = \frac{1}{R_1} + \frac{1}{R_2} \text{ parallel} \quad \left| \begin{array}{l} \\ \\ \end{array} \right. + \frac{\frac{1}{A_1 F_{12}} \left(\frac{1}{A_1 F_{13}} + \frac{1}{A_2 F_{23}} \right)}{\frac{1}{A_1 F_{12}} + \frac{1}{A_1 F_{13}} + \frac{1}{A_2 F_{23}}}$$

$$\frac{1}{R} = \frac{R_1 + R_2}{R_1 R_2} \quad \left| \begin{array}{l} \\ \\ \end{array} \right.$$

$$\text{Areas : } A_1 = A_2 = \frac{\pi}{4} \times (1.2)^2 = 1.131 \text{ m}^2$$

$$\varepsilon_1 = 0.81, \quad \varepsilon_2 = 0.63 \quad \text{given.}$$

View factor $F_{12} = 0.25$ given.

$$\text{Thus } F_{13} = 1 - F_{12} = 0.75 = F_{23} \text{ since } A_1 = A_2$$

$$\text{Then } R_T = \frac{1 - 0.81}{0.81 \times 1.131} + \frac{1 - 0.63}{0.63 \times 1.131}$$

$$+ \frac{1}{1.131 \times 0.25} \left(\frac{1}{1.131 \times 0.75} + \frac{1}{1.131 \times 0.75} \right)$$

$$\frac{1}{1.131 \times 0.25} + \frac{1}{1.131 \times 0.75} + \frac{1}{1.131 \times 0.75}$$

$$= 0.2074 + 0.5193 + \frac{3.537 \left(\frac{2.358}{1.179 + 1.179} \right)}{3.537 + 1.179 + 1.179}$$

$$= 0.7267 + 1.415 = 2.1415 \text{ m}^{-2}.$$

$$Q = 9400 \times 1.131 = 10631.4 \text{ W}$$

$$E_{b1} = E_{b2} + \alpha R_T$$

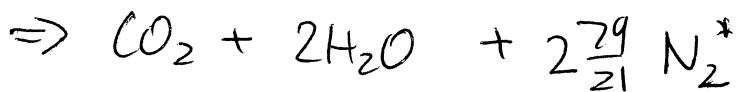
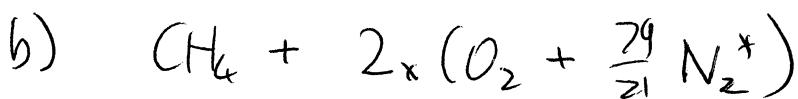
$$\sigma T_1^4 = \sigma T_2^4 + \alpha R_T$$

$$T_1^4 = T_2^4 + \alpha R_T / \sigma$$

$$T_1^4 = 500^4 + \frac{10631.4 \times 2.1415}{5.67 \times 10^{-8}}$$

$$4.015 \times 10^{-11}$$

$$T_1 = 825 \text{ K.}$$



$$\begin{aligned} \text{SFEE} \quad \dot{Q} &= H_{\text{PT}} - H_{\text{P298}} \leftarrow \left\{ 25^\circ \text{ Reactants} \right. \\ &= H_{\text{PT}} - H_{\text{P298}} - \Delta H_{\text{298}} - \left(\cancel{H_{\text{P298}}}^{\circ} - \cancel{H_{\text{P298}}} \right) \end{aligned}$$

$\Delta H_{\text{298}} \equiv -(\text{LCV})$ assuming H_2O in products
is vapour.

LCV (Haywood) = 50 000 kJ/kg_{CH₄}

Convert to moles = 800160 kJ/kmol_{CH₄}

Heat lost = 600kW (given)

Mass of CH₄ supplied = 85kg/hour = 0.0236kg/s

Heat lost/mole of CH₄ = $\frac{600}{0.0236} \times 16 = 406780 \frac{\text{kJ}}{\text{kmol CH}_4}$

$$\begin{aligned} \therefore H_{\text{PT}} - H_{\text{P298}} &= 800160 - 406780 \\ &= 393380 \text{ kJ/kmol}_{\text{CH}_4} \end{aligned}$$

Products

CO_2 H_2O N_2^*

moles 1 2 7.524

$\begin{cases} \text{H}_\text{mol} \\ 298 \end{cases}$ 9.37 9.90 8.67

$\begin{cases} \text{H}_\text{mol} \\ \text{total} \end{cases}$ 9.37 19.80 65.23 94.47 J

$\begin{cases} \text{H}_\text{mol} \\ 1500 \end{cases}$ 71.13 58.05 47.09

$\begin{cases} \text{H}_\text{mol} \\ \text{total} \end{cases}$ 71.13 116.1 354.3 $\frac{541.53 \text{ J}}{447.13 \text{ J}}$

$\begin{cases} \text{H}_\text{mol} \\ 1400 \end{cases}$ 65.31 53.39 43.62

$\begin{cases} \text{H}_\text{mol} \\ \text{total} \end{cases}$ 65.31 106.78 328.20 $\frac{500.29 \text{ J}}{405.89 \text{ J}}$
close, but high

$\begin{cases} \text{H}_\text{mol} \\ 1300 \end{cases}$ 59.55 48.84 40.19

$\begin{cases} \text{H}_\text{mol} \\ \text{total} \end{cases}$ 59.55 97.68 302.39 $\frac{459.62 \text{ J}}{365.22 \text{ J}}$

too small

Interpolate: $T = \left[\frac{1400 - 1300}{405.89 - 365.22} \right] \times (393.38 - 365.22) + 1300$

$$T = 1369^\circ\text{C}$$