

1) (a) (i)

The flows created when the pressure force is balanced by Coriolis force, are called Geostrophic flows. These flows are governed by

$$\frac{1}{g} \nabla p = - 2 \underline{\Omega} \times \underline{u}$$

where  $p$  includes the contributions from hydrostatic pressure and the centrifugal force.

The boundary layer between a geostrophic flow and a solid boundary with no-slip condition is called Ekman layer. The flow within this layer is known as Ekman layer flow and involves a balance among viscous, pressure and Coriolis forces. The governing equation is written as

$$\nu \nabla^2 \underline{u} - \frac{1}{g} \nabla p = 2 \underline{\Omega} \times \underline{u}$$

(ii) Density driven flows:

The density difference in the atmosphere can create flows due to buoyancy effects and these flows are known as density driven flows.

Examples are plumes, gravity currents, large scale motion in the atmosphere due to thermal or density stratification, avalanches.

(b) From the data Card:

$$\frac{Dq^2}{Dt} = - \overline{u_i u_k} \frac{\partial \bar{u}_i}{\partial x_k} \rightarrow \left( \frac{\partial \bar{u}_i}{\partial x_k} + \frac{\partial \bar{u}_k}{\partial x_i} \right) \frac{\partial \bar{u}_i}{\partial x_k} \\ + \frac{f_i u_i}{g} + \text{transport terms.}$$

Sources:

- (1) Wind shear, (2) Buoyancy effects

Sinks:

- (1) Viscous dissipation (2) Buoyancy effects.

Buoyancy effects can act as a sink or a source depending on the thermal or density stratification.

Sunny day  $\Rightarrow$  thermal effects,  $\frac{f_i u_i}{g} = \frac{g}{T} \frac{\partial \bar{u}_i}{\partial z} = \frac{g}{T} k_o \left( \frac{\partial T}{\partial z} \right)_z$ ,  
 (from the data card)

Some wind  $\Rightarrow$  wind shear effects;

$$- \overline{u_i u_k} \left( \frac{\partial \bar{u}_i}{\partial x_k} \right) = k_m \left( \frac{\partial \bar{u}}{\partial z} \right)^2 \quad \text{from the data card.}$$

Compare these two effects

$$\frac{\text{Thermal}}{\text{Mechanical}} = \frac{\frac{g}{T} k_o \left( \frac{\partial T}{\partial z} \right)_z}{k_m \left( \frac{\partial \bar{u}}{\partial z} \right)^2}$$

This ratio is known as Richardson's number.

$$\Rightarrow R_i = \frac{g}{\tau} \frac{k_0}{K_m} \frac{(\partial T / \partial z)}{(\partial u / \partial z)^2}$$

If  $R_i = 0$ ; No thermal stratification - Neutrally Stable.

$(\frac{\partial T}{\partial z}) < 0 \Rightarrow$  Unstable stratification  $\Rightarrow R_i < 0$  unstable.

$(\frac{\partial T}{\partial z}) > 0 \Rightarrow$  Stable stratification  $\Rightarrow R_i > 0$  stable.

(c) (i) Synoptic Scale (of the order of  $10^3$  km)

(1) Thermal effects

$\Delta T$  between pole & Equator

$\Delta T$  between Land & Ocean.

(2) Rotation of earth.

(ii) Meso Scale (of the order of  $10 - 10^2$  km)

(1) Land and Sea breeze

(2) Topography

(3) Heat island - cities.

(iii) Micro-scale (of the order of 1 m)

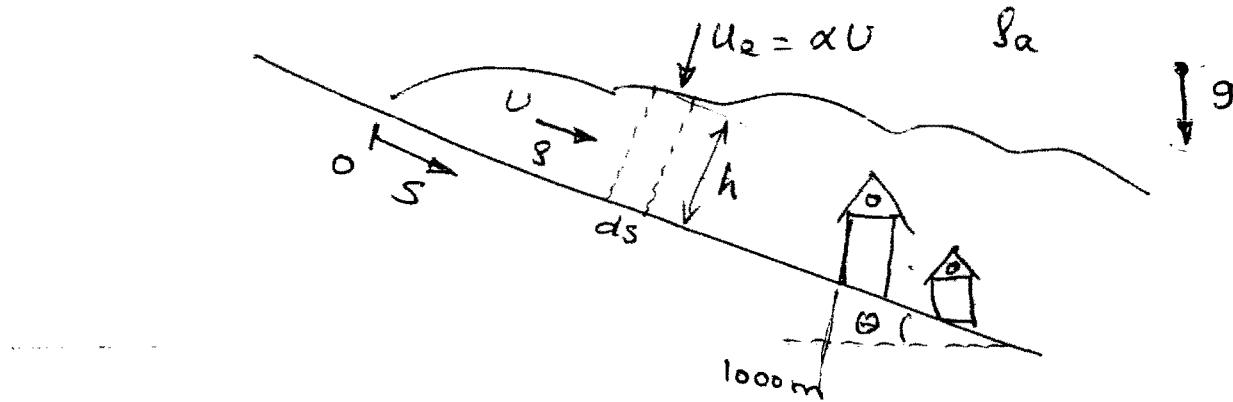
(1) Wind shear, (2) building wakes,

(3) local thermal effects.

Examiner's note:

The students showed a good understanding of geostrophic and Ekman layer flows, various sources and sinks for turbulence in the atmospheric flows. They were also able to deduce the stability conditions based on Richardson number for atmospheric flows.

(2)



$$\rho_a = 1.2 \text{ kg/m}^3, \quad \rho = 1.8 \text{ kg/m}^3$$

$$\alpha = 0.05$$

(a) Volume flux Conservation:

$$d(uh) = ds u_e \Rightarrow \boxed{\frac{d}{ds}(uh) = \alpha u} \quad - \textcircled{1}$$

mass conservation:

$$d(\rho uh) = \rho_a u_e ds \Rightarrow \boxed{\frac{d}{ds}(\rho uh) = \alpha \rho_a u} \quad - \textcircled{2}$$

Momentum Conservation:

$$d(\rho uh u) = ds h (\rho - \rho_a) g \sin\theta$$

$$\Rightarrow \boxed{\frac{d}{ds}(\rho u^2 h) = (\rho - \rho_a) g h \sin\theta} \quad - \textcircled{3}$$

using ① & ②

$$\frac{d}{ds}(\rho uh) = \rho_a \frac{d}{ds}(uh) \Rightarrow \boxed{\frac{d}{ds}[(\rho - \rho_a)uh] = 0} \quad \textcircled{4}$$

for Self-Similar Solution.

$$h = As^\alpha \quad u = Bs^\beta \quad (\rho - \rho_a) = Cs^\gamma$$

From ④  $\alpha = 1$ .

$$\text{From } ③ \frac{d}{ds} [AB^2 s^{a+2b}] = Agc s^{a+c}$$

$$\Rightarrow a+2b-1 = a+c \Rightarrow c = 2b-1$$

$$\text{From } ④ \frac{d}{ds} [ACB s^{a+b+c}] = 0$$

$$\Rightarrow a+b+c = 0 \Rightarrow a+b+2b-1 = 0 \Rightarrow b=0$$

$$\therefore c = -1$$

Note  $h = AS; U = B; (g - g_a) = c s^{-1}$

using ① again, one finds  $A = \alpha$ .

$$\therefore h = \alpha s, \quad U = B$$

(b) The village is at 1000 m from the source.

$$\Rightarrow h = 0.05 \times 1000 = 50 \text{ m} \cancel{\parallel}$$

(c)  $U = \text{const.}$  at  $s = 10 \text{ m}; h = 1 \text{ m}$   
 $\rho = 1.8 \text{ kg/m}^3$

$$U = 10 \text{ m/s}$$

The mass flow rate of the gas at this location

$$= \dot{m} = \rho U A = 1.8 \times 10 \times (1 \times 1) = 18 \text{ kg/s}$$

total volume flow at  $s = 10 \text{ m}$

$$\dot{V}_{10} = UA = 10 \times 1 \times 1 = 10 \text{ m}^3/\text{s} \text{ per unit width}$$

$$= \dot{V}_{10,a} + \dot{V}_{10,g}$$

at this location there is no air in the layer

(6)

$$\text{Since } \delta = 1.8 \text{ kg/m}^3$$

$$\Rightarrow \boxed{\dot{V}_{10} = \dot{V}_{10,g}}$$

Total volume flow at 1000 m

$$\dot{V}_{1000} = U \times A = 10 \times 50 \times 1 = 500 \text{ m}^3/\text{s} \text{ per unit width}$$

$$= \dot{V}_{1000,a} + \dot{V}_{1000,g}$$

$$= \dot{V}_{1000,a} + \dot{V}_{10,g} \quad (\text{Since volume flow of the gas is to be conserved})$$

$$\Rightarrow \dot{V}_{1000} - \dot{V}_{10} = \dot{V}_{1000,a}$$

$\therefore$  The rate of entrained air volume is

$$\dot{V}_{1000,a} = (500 - 10) = 490 \text{ kg/s per unit width.}$$

$$\therefore \text{mass of air entrained} = 1.2 \times 490 = 588 \text{ kg/s}$$

per unit width

mass fraction of the gas is

$$Y_g = \frac{m_g}{(m_g + m_{air})} = \frac{18}{18 + 588} = 0.0297.$$

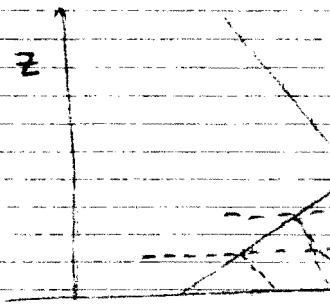
$$\boxed{Y_g = 0.0297}$$

Examiner's note:

The students were able to setup the required governing equations and deduce similarity solutions. Many students could not recognize that the volumetric rate of the entrained flow between two stations was given by the difference in the volumetric flow rates and also the chemical gas flow rate did not change with distance.

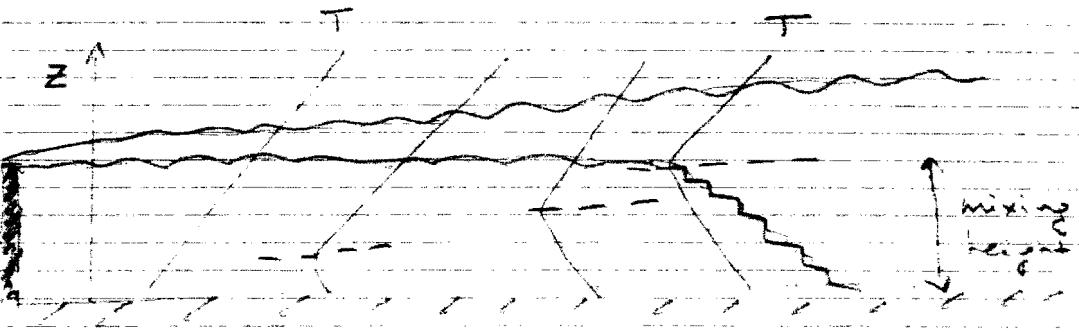
Q3

(a)



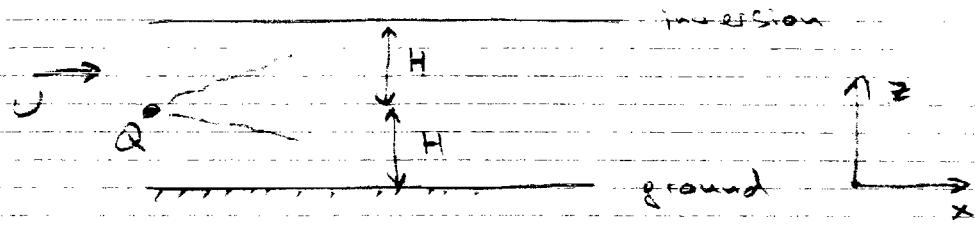
At night, radiative cooling of the surface implies colder air close to the ground than above it, resulting in

$T$ , a stable stratification. At dawn, the surface begins to warm up, giving a small reversal layer up to a height called "mixing height". In this region, turbulence is not damped and therefore pollutants may be well-mixed.



Answers: Pollutants emitted at night may travel a long distance undiluted if it is a stable atmosphere. At early dawn, as air warms height instead; it can't stabilize and the ground and air cool so that large concentrations of pollutants near the surface will diffuse. This is the reason for the early morning smog some may experience in autumn.

(b)



In the absence of ground and inversion,

$$\Phi(x, z) = \frac{Q}{\pi} \frac{1}{2\sigma} \exp\left(-\frac{(z-H)^2}{2\sigma^2}\right)$$

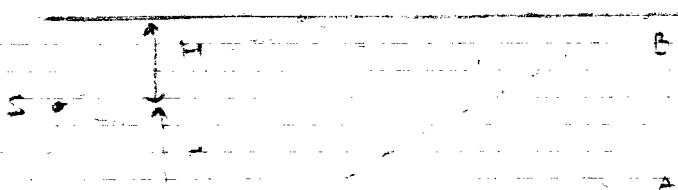
To take care of the ground, we need an image source  $S_1$  at  $z = -H$ . But, similarly, to take care

of the inversion, we need an image source at

$$z = 3H (S'_1). \quad S'_1 \rightarrow \text{source } \frac{\partial \Phi}{\partial z} \text{ at } z = -3H$$

 $S_2$ 

Due to  $S'_1$  (point A), we need  
an image source at  $z = -3H$   
( $S_2$ ) and similarly for point B,  
we need an image source  
at  $z = 5H$  ( $S'_2$ ).

 $S_1$  $S_2$ 

Ground effect  
inversion effect

$$z_1 = -H$$

$$z'_1 = 3H$$

$$z_2 = -3H \quad \text{and} \quad z'_2 = 5H$$

$$z_3 = -5H$$

$$z'_3 = 7H$$

so that the pollutant is given by

$$\Phi(x, z) = \frac{Q}{U \sqrt{2\pi} \sigma} \left[ \exp\left(-\frac{(z-H)^2}{2\sigma^2}\right) + \exp\left(-\frac{(z+H)^2}{2\sigma^2}\right) \right. \\ \left. + \exp\left(-\frac{(z+3H)^2}{2\sigma^2}\right) \right. \\ \left. + \dots \right]$$

$$+ \exp\left(-\frac{(z-3H)^2}{2\sigma^2}\right)$$

$$+ \exp\left(-\frac{(z-5H)^2}{2\sigma^2}\right)$$

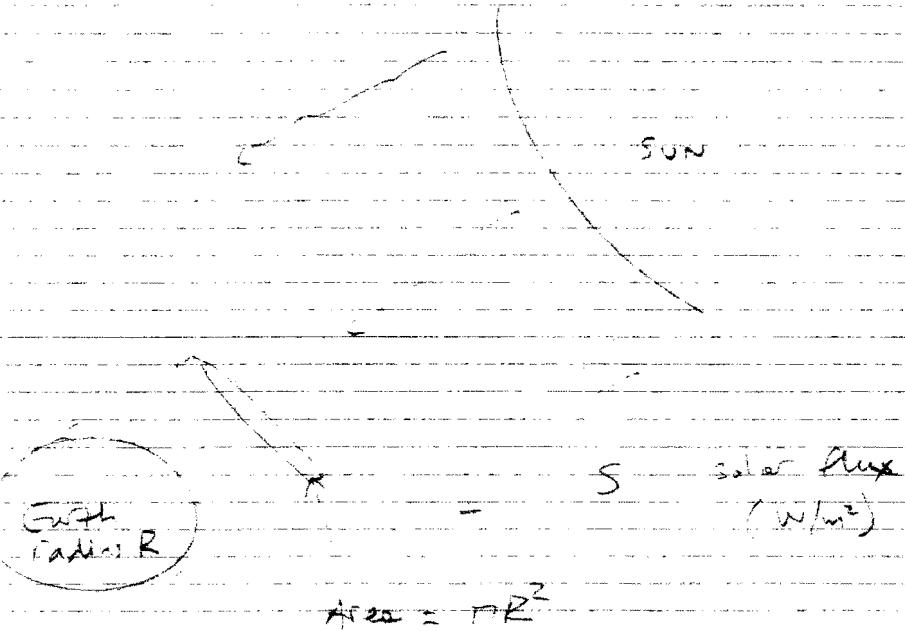
$$+ \dots$$

At close distance from the source, a few terms suffice, but at large distances many terms are needed.

### Examiner's note:

Few students clearly demonstrated the need for infinite number of image sources. Most candidates could describe the fumigation quite well.

Q4 (a)



$$\text{Earth吸收} \quad \pi R^2 S$$

$$\text{Reflected} \quad \alpha \pi R^2 S \quad (\text{Albedo}) \quad \text{Emissivity} (1-\alpha) = \epsilon S$$

$$\text{Emiss} \quad 4\epsilon S^2 \cdot 5 T_e^4 \quad \sigma = 5.67 \times 10^{-8} \text{ W/m}^2 \text{ K}^4$$

$$E = \sigma \epsilon S T_e^4$$

At earth's surface

$$(1-\alpha)S \pi R^2 = 4\epsilon S^2 \cdot 5 T_e^4$$

$$\Rightarrow T_e = \sqrt{\frac{(1-\alpha)}{4\epsilon}} R$$

So we get

eqn - derived in part b from which the first part of the question

is its consequence

so the final equation will be  $T_e = \sqrt{\frac{(1-\alpha)}{4\epsilon}} R$ and we will get  $T_e = \sqrt{\frac{(1-\alpha)}{4\epsilon}} R$ 

$$\Rightarrow \text{Final } T_e$$

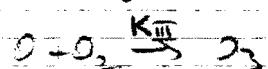
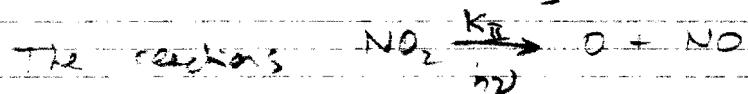
A more complicated model is needed to take fully into account  $\text{CO}_2$  and other greenhouse gases, but, approximately, presence of greenhouse gases affects  $\alpha$  &  $\epsilon$ . (11)

(b) "Smog" is a term to indicate photochemical pollution.

The primary pollutants  $\text{NO}$  &  $\text{VOC}$ 's produce

$\text{NO}_2$  and  $\text{O}_3$ , so above written after we have mainly  $\text{NO}$ ,  $\text{NO}_2$ ,  $\text{O}_3$  and  $\text{VOC}$ .

The reaction  $\text{NO} + \text{O}_3 \xrightarrow{K_1} \text{NO}_2 + \text{O}_2$  produces  $\text{NO}_2$ .



produce  $\text{O}_3$  -> first one is active and the presence of sunlight. The overall effect is to

produce high enough  $\text{O}_3$  and  $\text{NO}_2$  that a pollution episode may occur. The photochemical state is

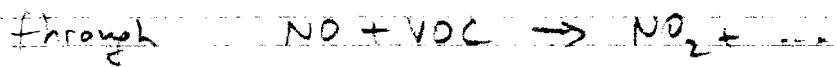
an approximation of the above  $\frac{d[\text{O}_3]}{dt} = 0$

$$\text{then } \frac{d[\text{O}_3]}{dt} = -k_1 [\text{NO}_2]_0 + k_2 [\text{NO}_2] = 0 \quad (\text{with } \frac{d[\text{O}_3]}{dt} = 0)$$

$$\Rightarrow \frac{d[\text{O}_3]}{dt} = \frac{k_2 [\text{NO}_2]}{k_1 [\text{NO}_2]}$$

for which can expect the following  
oxidative reaction -  $\text{O}_3^{\cdot}$

The presence of VOC also participates in the cycle (12)



hence effectively increasing  $\text{NO}_2$ .

Smog may also contain particulate matter that cause problems in their own right, but do not directly participate in photochemistry.

#### Examiner's note:

The qualitative description of pollution was given very well, but few students could describe the photo-stationary state deeply.